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UTILITY APPLICATION FOR UNITED STATES PATENT

FOR

DEVICE FOR DETERMINING THE AMOUNT OF SPLASH WATER TO WHICH A BRAKE PAD  
IS SUBJECTED ON A WET ROADWAY

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DEVICE FOR DETERMINING THE AMOUNT OF SPLASH WATER TO WHICH  
A BRAKE PAD IS SUBJECTED ON A WET ROADWAY

5       The present invention relates to a device for determining the amount of splash water, to which a brake pad is subjected on a wet roadway and to a usage of the device.

10       It is important for brake pads that they exhibit stable frictional values, even in wet conditions. Water in the form of spray water basically has a negative effect on the frictional values of a brake pad. It has transpired that with unfavourable constructions of the vehicle and/or brakes, the braking distance of a vehicle in winter can increase by a  
15       factor of up to 2 under the influence not only of water but also of road salt. This is of crucial significance for the safety of traffic. Furthermore, the comfort performance and the corrosion of the vehicle are negatively influenced by an unnecessarily high supply of sprayed water.

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      It is therefore the object of the present invention to provide a device of the type referred to above and a usage for this device.

25       In accordance with the invention, this object is solved by a device for determining the amount of splash water, to which a brake pad is subjected on a wet roadway, including a test brake pad, whose friction lining is hygroscopic such that it can absorb at least 5%, preferably at least 10%, water.

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      Substances which can withdraw the water vapour from a water vapour-gas mixture, eg. air, are termed hygroscopic.

Hygroscopic substances also absorb water when in direct contact with water, for instance when in contact with splash water in the invention. The water can be absorbed either by hydration, wherein water molecules are accumulated at polar groups of molecules dissolved in the water on the basis of electrostatic ion-dipole interaction or by forming hydrogen bridge compounds. The absorbed water can be gently removed again by the supply of energy. This is referred to as drying. The process of water absorption and drying occurs in certain temperature ranges without molecular conversion or splitting and is thus reversible in these ranges. The test brake pads in accordance with the invention can thus be used repeatedly, when using suitable test conditions, if they are dried sufficiently in the meantime. The material costs can be limited to a minimum in this manner.

Surprisingly, it has transpired that complicated test structures for determining the flow conditions of the spray water are not necessary. The vehicles can be tested in a test phase simply by replacing their standard brake pads by test brake pads. Since the friction lining in accordance with the invention is hygroscopic, there is a direct connection between the water absorption of the friction lining and the amount of moisture supplied to the friction lining in the form of spray water and air humidity. The test brake pad in accordance with the invention thus permits a precise determination of the moisture supplied to the brakes. The amount of spray water, to which a brake pad is subjected on a wet roadway, may be determined from this supplied moisture, either with the aid of comparative measurements or by means of a computer. In the simplest case, the influence of the air humidity can be ignored. In this manner, the information necessary for optimisation of the vehicle construction, particularly of the

underbody assembly and/or of the brakes as regards the supply of spray water can be obtained without particular effort and thus economically.

5       The above object is also solved in accordance with the invention by a use of the device in accordance with the invention, wherein

- a)    the initial mass of the test brake pad is determined,
- 10    b)   the test brake pad is installed in a vehicle,
- c)   the vehicle is subjected to predetermined operational conditions,
- d)   the final mass of the test brake pad is then determined and
- 15    e)   the water absorption of the friction lining is determined from the difference between the initial mass and the final mass.

Particularly high precision in the determination of the  
20 water absorption can be achieved if the test brake pad is dried before step d). Water adhering to the surface of the test brake pad may thus be removed so that it does not enter into the determination of the final mass of the test brake pad. The drying is effected such that the water absorbed by  
25 the test brake pad is not expelled. There is thus a brief drying process. It is particularly advantageous to wash off the test brake pad with a low boiling point liquid. Acetone or ethanol, above all, are suitable for this purpose. These liquids remove the surface water and the subsequent drying  
30 process then serves only to vaporise the washing liquid. The energy input necessary for this purpose is too low to expel the water absorbed by the test brake pad.

The steps a) to e) are advantageously performed for all brake pads of a vehicle. Problems with splash water generally occur more at the inner brake pads of vehicles than at the outer pads. As a result of the determination of the water  
5 absorption of all brake pads of a vehicle, the vehicle construction or the brake construction can be optimised substantially better, more rapidly and more economically.

A preferred exemplary embodiment is characterised in that  
10 a threshold value is determined for the water absorption and the construction of the vehicle and/or brakes is altered when the water absorption determined in step e) is greater than the threshold value.

15 The test brake pad in accordance with the invention is advantageously characterised in that the friction lining contains at least 15vol.%, preferably at least 17vol.%, hygroscopic bonding agent. The hygroscopic property of the friction lining can be achieved without difficulty by means of  
20 a substantially higher content of bonding agents in comparison to the prior art without special additives being required or mechanical treatment of the friction lining being necessary to achieve a porous body. The bonding agent can be selected at will in dependence on the desired manufacturing process, for  
25 instance with or without scorching. Resins (unmodified or organically or inorganically modified) are generally used as the bonding agent. Of the organic modifications, the cresol, alkyl, CSNL, epoxide, NBR, resorcinol, aryl, SBR and CR modifications have proved to be particularly satisfactory and  
30 of the inorganic modifications, the boron, phosphorus, silicone and chromium modifications have proved to be particularly satisfactory.

It has proved to be satisfactory if the friction lining is free of lubricant and, in particular, contains no sulphides or graphite. One embodiment of the invention is characterised in that the friction lining is additionally or alternatively free  
5 of abrasive agent and, in particular, contains no  $\text{Al}_2\text{O}_3$  , no Zr silicate and no SiC.

The term abrasive agents is to be understood in this case as abrasive frictional particles in the classical sense. It  
10 has transpired that a friction lining free of lubricant and free of abrasive agent is particularly suitable for testing the vehicle construction as regards the supply of splash water since it can not only be adjusted to be appropriately hygroscopic but also exhibits excellent frictional  
15 characteristics.

It is proposed in a further embodiment of the invention that the friction lining contains 8vol.% to 12vol.%, preferably 10vol.% fibres.

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The friction lining preferably contains aramide fibres and/or polyacrylonitrile fibres as the fibres. Aramide fibres, which already contain up to 5% water, can be used in order to reduce the electrostatic charge. These are able,  
25 under ideal conditions, permanently to store more than 15% water.

A preferred exemplary embodiment is characterised in that the friction lining contains

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6vol.% to 14vol.%, preferably 10vol.%, fibres,  
5vol.% to 13vol.% , preferably 9vol.%, rubber,  
13vol.% to 21vol.%, preferably 17vol.%, bonding agent,

10vol.% to 18vol.%, preferably 14vol.%, amorphous quartz,  
 1vol.% to 9.5vol.%, preferably 5.5vol.%, mica,  
 10.5vol.% to 18.5vol.%, preferable 14.5vol.%, magnesium-  
 aluminium silicate,

5 5.5vol.% to 13.5vol.%, preferably 9.5vol%, potassium  
 titanate,

6.5vol.% to 14.5vol.%, preferably 10.5vol.%, steel wool,  
 and

6vol.% to 14vol.%, preferably 10vol.%, aluminium  
 10 hydrosilicate

Both expanded and non-expanded mica can be used as the  
 mica. Potassium titanate can be included in the firm of  
 fibres or in powder form. The steel wool is for the purpose  
 15 of mechanical strength. It is inherently non-hygroscopic.

For test purposes, an appropriate friction lining was  
 stored at 95% air humidity. It transpired that the material  
 absorbed 3.4% water after only 2 days, 5.8% after 5 days,  
 20 10.2% after 14 days, 13.4% after 35 days and 15% after 49  
 days. This slow, continuous increase in the water absorption  
 is optimal for detecting the amount of spray water to which a  
 brake pad is subjected on a wet roadway, since the friction  
 pad can not immediately achieve its maximum water absorption  
 25 in the test phase, when it comes into contact with water. On  
 the other hand, the water absorption should not occur too  
 slowly since the test phase should not exceed reasonable time  
 limits. This is possible without difficulty with the  
 aforementioned composition. Furthermore, a test brake pad  
 30 with a friction pad of this composition exhibits stable  
 frictional values of between  $\mu=0.3$  and  $0.4$  under frictional  
 testing. The cold frictional value of  $\mu=0.4$  is also very  
 favourable and higher than with many mass produced materials.

Even emergency braking from 100 and 180km/h is stable. It is notable that such a test brake pad also has astonishingly short moment curves. Only at the first temperature fading, namely when the final temperature rises to 600°C, does the frictional value drop significantly. Overall, a test brake pad with a friction pad of the aforementioned composition has frictional characteristics comparable to the known, less hygroscopic, mass produced materials under temperature loading of up to ca. 400°C. Under the braking procedures performed in the test phase, account need consequently not be taken of the particular material characteristics of the test brake pad which is used. The operational conditions, to which the vehicle is subjected in the test phase, can therefore be selected to be as realistic as possible.

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A further advantage of the aforementioned composition is that a corresponding test brake pad results in no perceptible noise and, in particular, has a very good squeaking behaviour.

20 A hygroscopic rubber is preferably used as the rubber for reasons of compressibility. The friction pad preferably contains acrylonitrile - butadiene rubber as the rubber.

The automobile manufacturers and/or brake manufacturers can develop remedial measures to reduce the supply of splash water on the basis of the amount of splash water, to which a brake pad is subjected on a wet roadway and which may be detected with the test brake pad in accordance with the invention. They can alter the construction of the vehicle or of the brakes or incorporate additional elements, for instance protective shields or deflecting spoilers. The braking distance can be shortened under wet conditions in this manner and the safety of the vehicle and thus of the traffic can be



significantly increased. At the same time, corrosion is reduced. Numerous developments are possible within the scope of the inventive concept. The friction lining can be as strongly hygroscopic as desired. The composition of the  
5 friction pad can also be altered as desired, provided that a minimum amount of frictional properties remains. Instead of determining the initial mass and the final mass of the test brake pad, the corresponding initial volume and final volume can also be determined.